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11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter coefficients subscripts mass

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SECTION 11-1 REVIEW AND REINFORCE The Water Cycle ^ Understanding Main ideas Study the illustration, and then answer the questions on a separate sheet of paper. 1. What three processes does this illustration show at points A, B, and C?ri'f ' 2. What is the source of energy that drives the water cycle? I lit: 11 3.

SECTION 11-1 REVIEW AND REINFORCE

11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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2 AnswerKey 2-1 Explore, page 5 2-1 Review and Reinforcement, pages 8-9 1 energy 2 radiant 3 chemical potential 4 joule 5 created, destroyed 6 natural gas 7 kinetic 12 filtration 13 crystallization 14 distillation 15 chromatography 16 electrolysis arXiv:1701.07274v6 [cs.LG] 26 Nov 2018

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11—1 Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid? The equation for this reaction is $Mg + 2HCl \rightarrow MgCl_2 + H_2$ 14. Aluminum reacts with HCl to produce aluminum chloride ($AlCl_3$) and hydrogen ...

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24—1 Review and Reinforcement (continued) 13. There are 3.29 g of iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days. How old is the sample? 14. A sample that originally contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is 6×10^{10} years.

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